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- (i) Nitration of benzene
- (ii) Halogenation of benzene
- (b) Why Friedel-Craft acylation is preferred over Friedel-Craft alkylation?
- (c) Why nitration of toluene is easier than benzene? 2



SECTION - D

- 8. (a) Explain two methods of formation of alkyl
- (b) Describe the addition elimination mechanism of nucleophilic aromatic substitution.
- 0 Why allyl halides are more reactive while vinyl halides are less reactive than alkyl halides?
- 9 (a) Explain:

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- (i) Gattermann reaction
- (ii) Hunsdiecker reaction
- (b) Explain the mechanism and stereochemistry of SN1 reaction with energy profile diagram.
- (c) Write the preparation of DDT and BHC.

B. Sc. (Hons.) Chemistry 2nd Sem.

91535

Latest Examination - April, 2018

ORGANIC CHEMISTRY

Paper: 203

[Maximum Marks: 40

Time: Three Hours]

have been supplied the correct and complete question paper. No Before answering the questions, candidates should ensure that they complaint in this regard, will be entertained after examination.

Note: Attempt five questions in all, selecting one question from each Section. Question No. 1 is compulsory. All questions carry equal marks

- 1. (a) What is reforming or aromatization?
- (b) Why trans alkenes have zero dipole moment?
- (c) What are dienophiles? Give examples.
- (d) What happens when ethyne is treated with HCN?
- (e) Why benzene undergoes substitution rather than addition reactions?
- What is Fittig reaction?

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- (g) Why nitrobenzene is used as a solvent in Friedel-Craft reactions?
- What is Swarts reaction?

 $1 \times 8 = 8$

SECTION - A

- 2. (a) Why cis-alkenes have higher boiling point than corresponding trans alkenes?
- (b) Dehydration of alcohols to alkenes is carried out by heating with H2SO4 and not with HNO3 and HCl. Explain.
- Explain hydroboration-oxidation reactions alkenes with mechanism. of
- 3. (a) Describe:
- (i) Peroxide effect
- (ii) Hofmann elimination reaction
- (b) Write note on :

4

- (i) Cetane number
- (ii) Cracking

SECTION - B

What are conjugated dienes? Why they are more stable than isolated dienes?

- (b) Why terminal alkynes are acidic in nature?
- (c) Describe the structure and bonding in alkynes.
- (a) Illustrate the Pschorr synthesis of phenanthrene. 2
- (b) How does anthracene reacts with the following reagents:
- (i) Maleic anhydride
- (ii) Chromic acid
- (c) Describe 1, 2 and 1, 4-addition reactions of conjugated dienes.

SECTION - C

- (a) Classify the following into aromatic, anti-aromatic and non-aromatic compounds:
- (i) Cyclo-octatetraene
- (ii) Pyrrole
- (iii) Cyclopropenyl anion
- (iv) Cyclobutadiene
- (b) Explain:

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- (i) Huckel rule of aromaticity
- (ii) Annulenes

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