

Roll No.

- (a) What is magnetic moment ? Also discuss orbital contribution to magnetic moment of substances for high spin and low spin complexes. 8

- (b) Draw and discuss structure of $\text{B}_{10}\text{H}_{14}$ and meta- $\text{C}_2\text{B}_2\text{H}_{12}$. 8

SECTION - D

- (a) Explain bonding in carbonyl complexes. 6
- (b) Discuss the structure of sodium nitroprusside. 5
- (c) Write short note on tertiary phosphine as ligand. 5
- (d) Describe linear and bent Nitrosyl Ligand with suitable examples. 8
- (e) Why dinitrogen is weaker ligand in comparison to carbonyl ligands ? Explain by M.O. diagrams of each. 8

Time : Three hours / Maximum Marks : 80

Before answering the questions, candidates should ensure that they have been supplied the correct and complete question paper. No complaint in this regard, will be entertained after examination.

- Note :** Question No. 1 is **compulsory** and all questions carry equal marks. Attempt **five** questions in all, selecting at least **one** question from each Section.
- 1. Compulsory Question :** 2+2+2+2+2+2=16

- (a) What are homonuclear and heteronuclear carbonyls ?
- (b) What is Nephelauxetic effect ?
- (c) Explain Synergic effect.

P. T. O.

M. Sc. (Chemistry) 2nd Semester

Examination – May, 2019

INORGANIC CHEMISTRY-II

Paper : CY(H)-201

SECTION – B

- (d) What is Neil's temperature ?
- (e) What is G.S.T. for Mn^{2+} ion ?
- (f) Predict structure of $[Ru_5N(CO)_4]$ using skeletal and non-skeletal electrons.
- (g) Differentiate between atomic orbital and molecular orbital.
- (h) Define T.I.P.
- 4.** (a) Discuss electronic spectra for molecular compounds of I_2 .
- (b) Discuss Orgel diagrams for d^2 and d^4 for octahedral complexes. Also predict possible transitions.
- 5.** (a) Discuss and draw the T-S diagram for d^5 systems.
- (b) Explain LMCT transitions in tetrahedral complexes taking suitable examples.
- (c) Explain John Tellor distortion with suitable examples.

SECTION – A

- 2.** (a) Draw and explain M.O. diagram for the octahedral complex $[Cr(H_2O)_6]^{3+}$. 8
- (b) Discuss all the factors influencing the stability of coordination compounds. 8
- 3.** (a) Discuss Crystal Field Theory. 8
- (b) Discuss effect of π -overlapping on the magnitude of splitting energy, considering only π -overlapping with d_{xy} orbitals of metal used. 8
- 6.** (a) Explain Guoy's method for determination magnetic susceptibility.
- (b) What are L.N.C.C ? Also discuss structure and bonding in following :
 $(\mu-CO)_2(CpRh)_3CO$ and $n^4-(C_4H_4)_2Fe_2(CO)_3$

(2)

(3)