

**UNIT – IV***Roll No. ....*

- (a) Explain the rate process approach to ionic migration and derive an equation for the equivalent conductivity on the basis of it. 8
- (b) Derive the Nernst Planks Flux equation and Discuss its consequences. 8

- (a) Give an account of Debye-Hückel-Onsager theory of conductance of strong electrolytes. 8
- (b) Write a note on the Onsager Phenomenological equations. 8

**22002****M. Sc. Chemistry 2nd Semester****Examination – May, 2019****PHYSICAL CHEMISTRY-II****Paper : CY(H)-202***Time : Three hours / Maximum Marks : 80*

*Before answering the questions, candidates should ensure that they have been supplied the correct and complete question paper. No complaint in this regard, will be entertained after examination.*

**Note :** Attempt any *five* questions, selecting at least *one* from each Unit. Question No. 1 is *compulsory*. All questions carry equal marks.

1. (a) What is probability distribution function?  
 (b) What are spherical harmonics?  
 (c) What is competitive and non-competitive inhibition?  
 (d) What is Walden rule?

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- (e) What are the characteristics of chain length?
- (f) What is meant by steady state approximation?
- (g) What is an explosion reaction? Give example.
- (h) Explain the factors effecting ionic mobility.

$$2 \times 8 = 16$$

### UNIT - I

2. (a) Solve the Schrodinger wave equation for the energy of a particle in three.

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- (b) Dimensional box.

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3. Discuss the shape of p-orbital using the concept of quantum mechanics. Set up the Schrodinger wave equation for a Rigid Rotator and obtain equations for the energy eigen values and normalized eigen function.

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### UNIT - II

4. (a) Discuss the Third law of thermodynamics and its limitations.

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7. Derive Michaelis-Menton equation of ~~enzym~~  
catalyzed equation. Evaluate Michaelis constant  $K_m$   
Lineweaver - Burk and Eadie Hofstae methods.

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- (b) Discuss Clausius-Clayperon's equation application.

5. (a) Discuss phase diagram of two completely immiscible components system.

- (b) Discuss Phase Diagram of Eutectic ~~system~~  
congruent and incongruent melting point.

### UNIT - III

6. (a) Discuss the general treatment of chain ~~reactions~~  
and explain the concepts of apparent activation energy and chain length.

- (b) What are explosion reactions? Discuss by ~~using~~  
H<sub>2</sub>-O<sub>2</sub> example.

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